ADVANCEMENTS IN ISOTOPIC LABELING: H/D EXCHANGE ON *N*-HETEROCYCLES AND THE DEVELOPMENT OF A ¹³CO₂ GENERATOR

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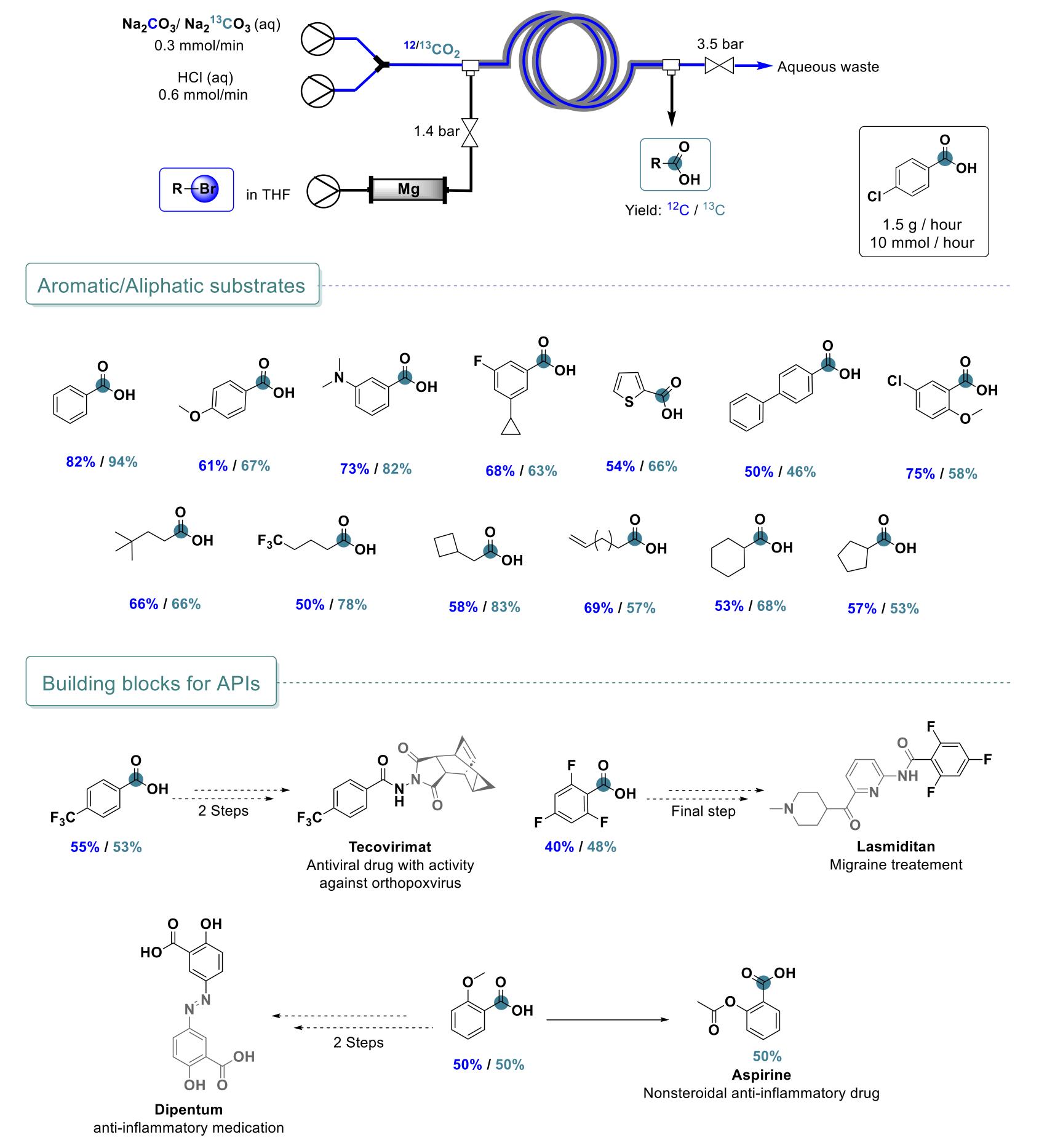
INTRODUCTION

Isotopically labeled compounds play a crucial role in material sciences, physical organic chemistry, and in pharmaceutical research, especially for the identification and quantification of metabolites during drug development. The most compelling approach for preparing isotopically labeled compounds is through direct isotope exchange, as it allows practitioners to perform the labelling without the need for pre-functionalization.

X-Chem is the pioneer of DNA-encoded library (DEL) technology and a leader in small molecule drug discovery. We combine proven DEL and medicinal chemistry expertise with groundbreaking AI/ML solutions and expert custom synthesis. We bring the most modern and powerful synthetic methodologies to bear, including asymmetric catalysis, photoredox, and flow chemistry. Herein, we present our latest results in the fields of H/D exchange and ¹³C labeling.

¹³CO₂ GENERATION AND LABELING

On-demand ¹³CO₂ generator: The setup consists of tube-in-tube reactor and utilizes Na₂¹³CO₃ as an easily available precursor. As a proof of concept, the on-demand generated ¹³CO₂ was reacted with Grignard intermediates to form ¹³C-labeled carboxylic acids.



H/D exchange. We present a continuous flow Raney nickel-catalyzed hydrogen isotope exchange process that boasts compatibility with a wide spectrum of nitrogen-containing heterocycles and pharmaceutical compounds.

¹³C labeling. We describe a practical and economical chemical generator for CO₂. When used with ¹³C-labeled Na₂CO₃, the generator provides alternative to expensive ¹³CO₂ cylinders.

HYDROGEN ISOTOPIC EXCHANGE (HIE)

- Labeling reactions were carried out in an H-Cube Pro[™] flow reactor.
- Five heterogenous catalysts were evaluated (10% Pt/C, 10% Pd/C, 2.5-2.5% Pd-Pt/C, 5% Ru/C, Raney nickel), with Raney nickel demonstrating the most optimal results.
- Flow reactions provided superior results compared to batch.

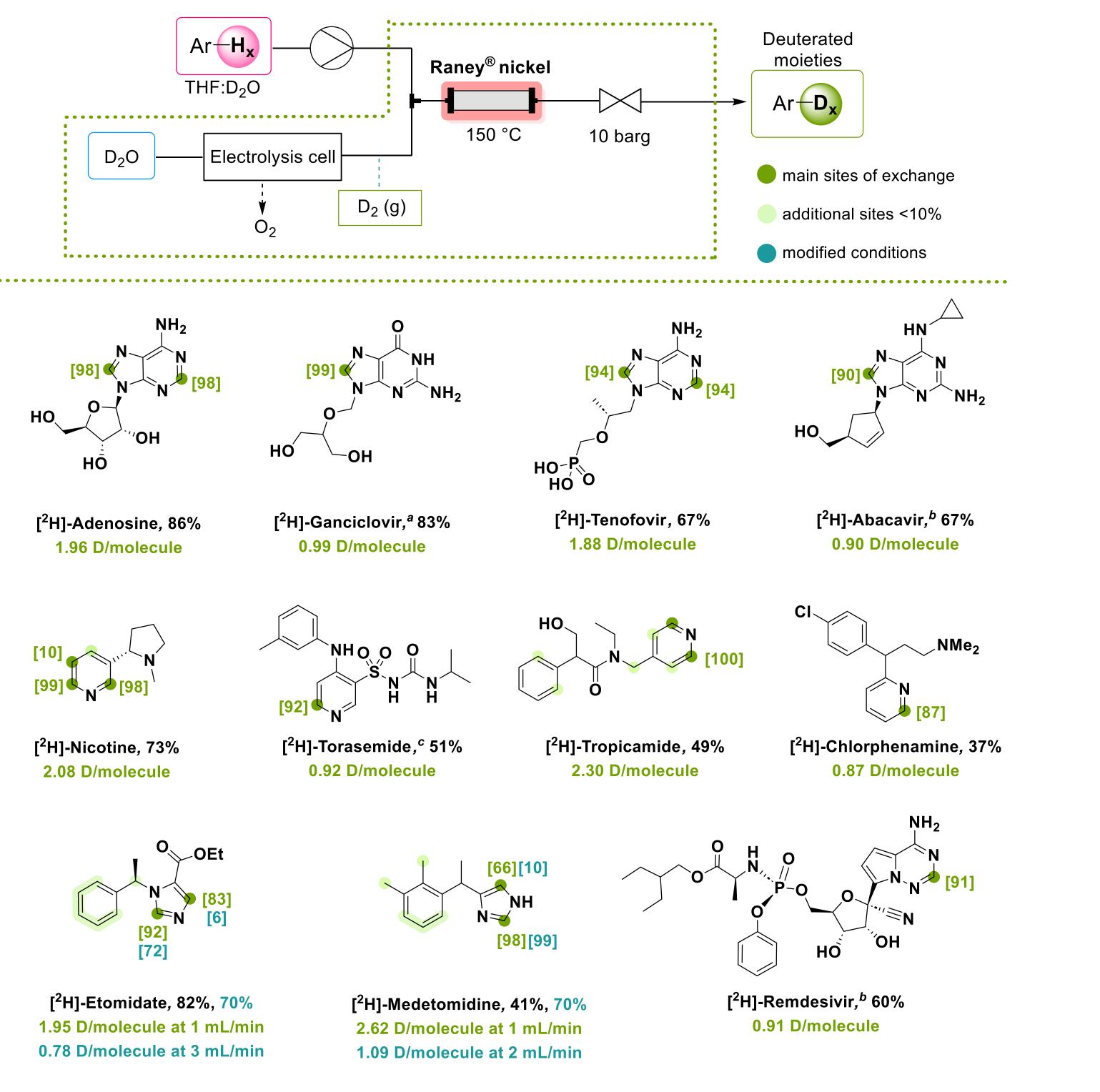
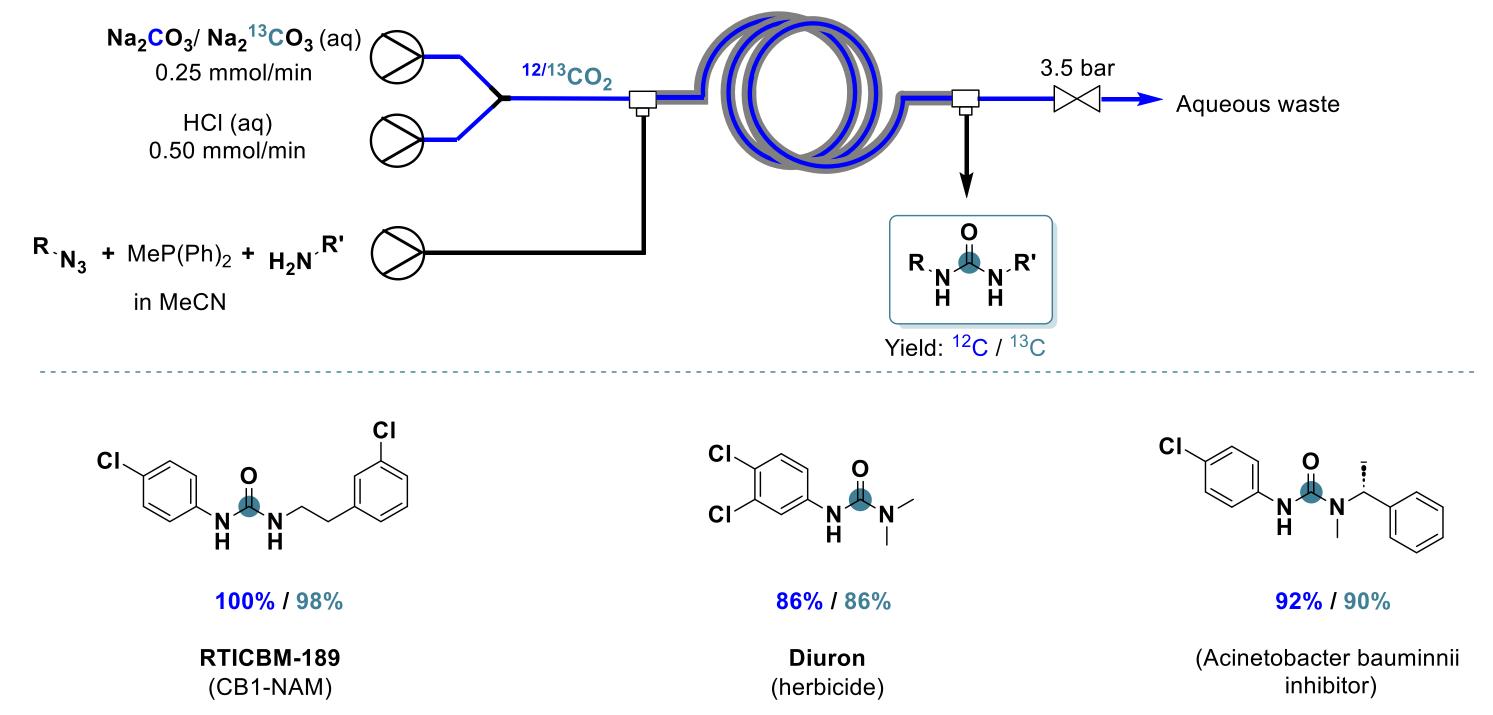


Figure 1. Labeling of APIs. Sites of deuterium incorporation were determined by ¹H NMR spectroscopy. General conditions: substrate (1 mmol) in THF:D₂O = 95:5 (0.01 M), 30 mm RaNi CatCart, 150 °C, 10 bar, 3% D₂, 0.3 - 2 mL/min. ^{*a*} 0.05 M; ^{*b*} without D₂; ^{*c*} 130 °C.

Figure 3. On-demand CO_2 generation. General conditions: substrate (3.5 mmol, 1 equiv), $Na_2^{12}CO_3/Na_2^{13}CO_3$ (0.8 equiv), THF (6 mL), room temperature, 15 min reaction time, 0.5 mL/min.

SYNTHESIS OF ¹³C-LABELED ASYMMETRIC UREAS

The CO₂ generator was also utilized for the synthesis of both ¹²C- and ¹³C-labeled biologically active asymmetric ureas through the Staudinger/aza-Wittig sequence.



GRAM SCALE HIE IN FLOW

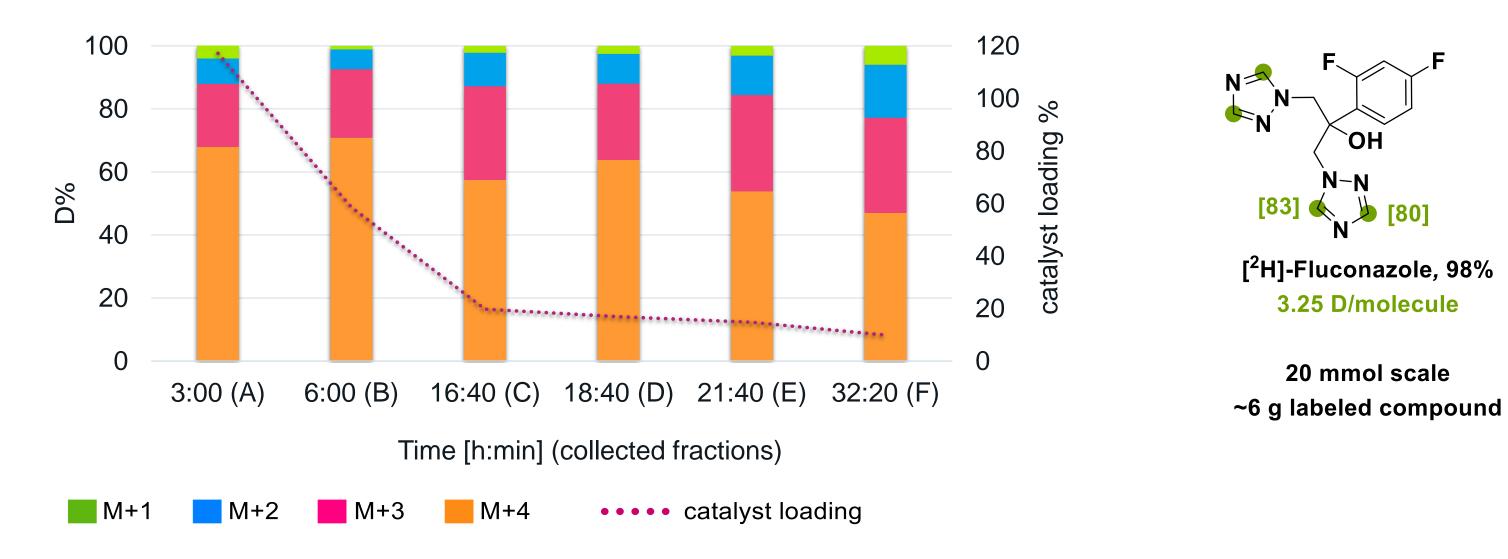


Figure 2. Gram scale synthesis of [²H]-Fluconazole

References: ¹ HIE: G. Sipos et al. ChemRxiv, **2024**, doi:10.26434/chemrxiv-2024-tk50j-v2; ² Chemical generators: Kappe et al. Acc. Chem. Res. **2020**, 53, 1330–1341; ³ Urea synthesis in flow: Audisio et al. Asian J. Org. Chem. **2023**, 12, e202200640

Figure 4. Synthesis of asymmetric urea compounds. General conditions: amine (0.3 mmol, 1 equiv), azide (0.3 mmol, 1 equiv), MeP(Ph)₂ (0.3 mmol, 1 equiv), Na₂¹²CO₃/Na₂¹³CO₃ (1.75 equiv), MeCN (3 mL), room temperature, 3 min reaction time, 750 μ L/min.

SUMMARY

- We described a continuous flow Raney nickel-catalyzed hydrogen isotope exchange process. The broad applicability of the method was demonstrated through successful labeling of various purine bases, imidazoles, pyridines, and active pharmaceutical ingredients.
- We developed a practical and economical CO₂ generator which utilizes Na₂CO₃ as a readily available precursor. The CO₂ generator was employed in the synthesis ¹³C-labeled carboxylic acids and non-symmetrical urea compounds.



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